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Stereospecific Synthesis of 9-Demethylretinoids via Palladium-Catalyzed Vinylboronic Acid-Vinyl Iodide Cross Coupling

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Abstract: A stereospecific synthesis of 9-demethylretinoids with either trans or 11-cis geometries, based on the thallium-accelerated palladium-catalyzed cross-coupling reactions of an (E)-1-alkenylboronic acid and (E) or (Z)-alkenyl **iodidea, is described.**

Artificial rhodopsins and bacteriorhodopsins.1 obtained by replacing the native chromophore **(1 1-cis-retinal** 1 in rhodopsin and trans-retinal 2 in the light-adapted form of bacteriorhodopsin) with a synthetic retinal analog have been instrumental in the present knowledge of the visual cycle and the bacteriorhodopsin photocycle. It has been recently shown that pigments derived from 11-cis-9-demethylretinal 3 fail to produce the biochemically active form of rhodopsin (meta II) upon photoisomerization.² This result led to the suggestion^{2a,b} that the C9methyl group in rhodopsin connects the isomerization of the retinal to later events in the visual cyqle by triggering a *cis-trans* isomerization of the peptide bond adjacent to the amino terminal side of a proline residue during the lumi-meta I transition, necessary for G-protein activation of the visual process. On the other hand, it is wellknown that trans isomers of side-chain demethylretinals such as 4 modulate the proton-pumping activity of bacteriorhodopsin.^{2c}

The dependence of biological profiles on the stereochemically defined structure of the polyene side chain makes it necessary to develop synthetic approaches to retinoids with high chemo-, regio- and stereoselectivities. Existing syntheses of mixtures of side-chain 11-cis- and trans-demethylretinoids rely on Wittig^{3a} and Horner-Emmons reactions.^{3b} Alternatively, isolation of the 11-cis congeners following photochemical isomerization^{3a} requires their separation from the more abundant *rrans* and other retinal isomers with cis-geometries about the most substituted double bonds. In this regard, coupling of stereochemically pure vinylic fragments seems a most reasonable approach, provided that the cross-coupling reaction takes place with retention of the stereochemistry. An in-depth study describing some palladium-catalyzed alkenyl-alkenyl coupling reactions for the synthesis of derivatized vitamin A has recently been published.⁴ In contrast to the efficient coupling between alkenylzinc and alkenyl iodides described in Negishi's paper4, alkenylboronic esters performed rather poorly under the classical conditions developed by Suzuki.5

We are reporting a stereo-defined approach to retinoids, based on the modified Suzuki cross-coupling reaction of an (E)- 1 -alkenylbomnic acid and stereochemically pure vinyl iodides catalyzed by palladium at mom temperature, which complements nicely the above study, and is of practical application to the preparation of the unstable and difficult-to-obtain 11 *-cis* isomers.

In a first approach to retinoids we prepared the known6 boronic acid 5 via direct treatment of Q-3-methyl-2-penten-4-yn-l-01 with catecholborane followed by hydrolysis. Although compound 5 has been used for the preparation of smaller fragments in the synthesis of side chain retinoid analogs,7 the low yields in the difficult separation of the catechol byproduct called for a change of functionality of the different coupling partners. At this point, we faced the problem of the lack of reactivity of dienynes such as 78a and trienynes such as 88b with catecholborane, even under forced conditions (starting material being recovered after prolonged heating in C_6D_6 as solvent, as proved by NMR monitoring). Fortunately, the catalytic effect⁹ of the borane-N,N-diethylaniline complex allowed the efficient preparation of vinylboronate 9 from dienyne 7 at room temperature (Scheme 1).¹⁰

Scheme 1. Reagents and reaction conditions *i.* 1. Catecholborane (1.1 equiv), botane-N,N-diethylaniline complex (0.11 equiv), benzene, rt, 9 h. *ii.* **H200. rt. 3 h; 91% overall yield.**

On the other hand, highly stereoselective routes to vinyl iodides (Scheme 2) relied upon well-known alkenylation reactions of aldehydes. (E)-Vinyl iodides (13 and 14a) were obtained from ester 11 or protected alcohol 12 by the **CHI3-CrCl2** reaction developed by **Nakail** 1. (Z)-Vinyl iodides (15 and 16) were prepared by Wittig reactions of the same substrates with iodomethylenetriphenylphosphorane as described independently by Stork^{12a} and Bestmann^{12b}. The major isomer in each case, obtained after careful chromatographic separation (appreciable amounts of the fruns isomers were obtained in the purification of the cis isomers if the silica gel was not pretreated with pyridine), was subjected to the coupling reaction with the boron-functionalized fragment.

Scheme 2. Reagents and reaction conditions

i. 1. ethylene glycol, PPTS, benzene, reflux, 12 h; 2. LiAiH₄, ether, 0 °C, 1 h; 3. TBDMSiCl, imidazol, DMF, rt, 1.5 h;

4. pTSA, acetone-H₂O, rt, 15 min. *ii. CrCl₂*, CHI₃, THF, 0 °C, 45 min. *iii.* (Ph₃P⁺CH₂I)I['], NaN(TMS)₂, THF, HMPA, -78 °C, 1 h.

For the coupling reaction we turned our attention to the TlOH modification of the Suzuki coupling developed by Kishi and coworkers for one of the critical steps in the synthesis of fully protected palytoxin carboxylic acid.13 Even under those conditions, vinylboronates performed poorly in accordance with the above observation.4 The boronic acid 10, obtained by hydrolysis of bomnate 9 (Scheme 1) gave reproducible coupling rates with the vinyl iodides used (probably due⁶⁶ to the better quality of the reagent utilized) and acceptable yields of 9-demethylretinoids. A representative experiment involves stirring of a degassed THF solution of boronic acid 10 (1.0 equiv) and vinyl iodide (13-16, 0.67 equiv) in the presence of 10% TlOH (3.08 equiv) and Pd(PPh₃)4 (0.07 equiv) under inert atmosphere for 30 min at rt. Under the described conditions, **1 1-cis** -9demethylretinoid.s 17 and 18 were obtained in 57% and 67% yields of purified product, respectively (Scheme 3). *Trons-9* demethylretinoids 19 and 20 were prepared in *48 %* and 31% yields. respectively, again with essential retention of configuration. both in (E)-1-alkenylboronic acid and in (E) or (Z)-alkenyl iodides. no other isomers being detected in the reaction **mixture.l4**

Scheme 3. Reagents and reaction conditions *i.* $(n-Bu)_{4}$ NF $(1M \text{ in } THF)$, $n \text{ , } 4 h$. *ii.* MnO₂, Cl₂CH₂, $n \text{ , } 2 h$.

Finally, fluoride-induced deprotection of 17 and 19 to the 9-demethylretinols (21 and 22) followed by MnO₂ oxidation afforded the known 11-cis-9-demethylretinal 3 and trans-9-demethylretinal 4, respectively (Scheme 3) whose spectroscopic data coincide with those published.^{3a} A direct route to 11-cis-9-demethylretinol **21(50 96** overall yield) uses the direct coupling of boronic acid **10** and alcohol **14b,** itself obtained (Scheme 2) by **DIBAL-H reduction** of ester **13** (alternative preparation of **14b** by fluoride deprotection of **14a** led to some isomerization).

We believe that the mild conditions of the cross-coupling reaction (compatible with carboxylic esters and unprotected alcohols) make this procedure particularly useful for the highly efficient stereo- and tegiospecific preparation of retinoids, especially the difficult-to-obtain **1 1-cis** isomers, for their use in a variety of biological protocols. Extension of this methodology to the stereospecific synthesis of the parent system as well as a variety of retinoids with structural modifications at the cyclohexenyl ring will be reported in due course.

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References

- 1. (a) Dawson, M. I.; Okamura, W. H. (eds.); *"Chemistry and Biology of Synthetic Retinoids",* CRC Press, Boca Raton, FL. 1990. (b) Parker, L.(ed.) *"Methods in Enzymology",* vol. *89* Retinoids Part A, Academic Press, N.Y., 1990. (c) Parker, L.(ed.) 'Merhods *in Enzymology",* vol. 90 Retinoids Part B, Academic Press, N.Y., 1991. (d) Spom, M. B.; Roberts, A. B.; Goodman. D. S. (eds.); *'The Rednoids":* Academic Press, New York, 1984; volumes 1 and **2.**
- 2. (a) Ganter, U. M.; Schmid. E. D.; P6rez-Sala, D.; Rando, R. R.; Siebert, F. *Biochemisrty 1989,28, 5954.* (b) Caiiada, F. J.; Law, W. C.; Rando, R. R.; Yamamoto, T.; Derguini. F.; Nakanishi, K. *Biochemistry 1990,29,9690. (c)* Giirtner, W.; Towner, P.; Hopf, H.; Oesterhelt, D. *Biochemistry 1983, 22, 2637.*
- 3. (a) Broek, A. D.; Muradin-Szweykowska, M.; Courtin, J. M. L.; Lugtenburg, J. *Reel. Trav. Chim. Pays-Bas 1983, 102, 46. (b) van den Tempel, P. J.; Huisman, H. O. Tetrahedron 1966, 22, 293.*
- $\overline{\mathbf{4}}$. Negishi, E.; Owczarczyk, 2. *Terrahedron Left.* **1991,32,6683.**
- $5.$ (a) Miyaura, N.; Yamada, K.; Suzuki, A. *Tetrahedron Let?. 1979,3437.* (b) Miyaura, N.; Suginome, H.; Suzuki, A. *Tetrahedron Lert. 1981.22, 127. (c)* Miyaura, N.; Yamada, K.; Suginome, H.; Suzuki, A. *J. Am. Chem. Sot.* **1985,107, 972.**
- 6. **(a) Roush, W. R.;** Brown, B. B.; Drozda, S. E. *Tetrahedron Z&r. 1988,29, 3541.* (b) Roush, W. R.; Moriarty, K. J.; Brown. B. B. *Terrahedron L&t. 1990.31.6509.*
- 7. Unpublished results from these laboratories.
- 8. (a) Negishi, E.; King, A. 0.; Klima, W. L.; Patterson, W.; Silveira, A. *J. Org. Chem. 1980,45, 2526. See* also Negishi, E.; King, A. 0.; Tour, J. M. *Org.* Synth. 1985.64, 44. (b) Matsumoto, M.; Kuroda, K. *Tetrahedron Lett. 1980,21,4021.*
- 9. Suseela, Y.; Prasad, A. S. B.; Periasamy, M. *J. Chem. Sot., Chem. Commun.* **1990,446.**
- 10. *The* effect of the borane-N,N-diethylaniline complex, that is the hydrobotation and exchange of the alkenyl/alkyl group with a diaryloxyborane such as catechol borane, was not observed with trienyne 8. Compound **10** is, to our knowledge, the only reported example of a conjugated trienylboronic acid.
- 11. Takai. K.; Nitta, K.; Utimoto, K. *J. Am. Chem. Sot. 1986,108,7408.* It is reported in this article that α , β -unsaturated aldehydes give a mixture of the four diastereoisomers of the δ -iododienal.
- 12. (a) Stork, G.; Zhao, K. *Terruhedron Left. 1989,30,2173.* (b) Bestmann, H. J.; Rippel, H. C.; Dostalek, R. Tetrahedron Lett. 1989, 30, 5261.
- 13. Uenishi. J.; Beau, J.-M.; Armstrong, R. W.; Kishi, Y. *J. Am. Chem. Sot. 1987,109. 4756.*
- 14. Satisfactory spectroscopic data (tH NMR, I3C NMR, IR, UV) was obtained for all the new substances reported in this paper. Their elemental composition was determined by high resolution mass spectroscopy.

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